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called acidic; if the pH is about 7, the solution is neutral; if the pH is greater than 7, the solution is called basic. In an acidic solution, then, the concentration of hydrogen ions is greater than the

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Indicators are used to determine whether a solution is acidic or alkaline. Acids react with metals, bases and carbonates to produce salts. Neutralisation is the reaction

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between an acid and a base.

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