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Note-Taking-Guide-Episode-304-Name-Quantum-Numbers

Note Taking Guide: Episode 304 Name Aditi Acharya Quantum Numbers N, l, m, s Used to describe an electron in an atom n Principle quantum number Represents main energy level of electron maximum # of electrons in an Energy level = 2n² Example: What is the maximum number of electrons that can be in the 5th main energy level? l The 2 nd quantum number • Describes the orbital shape within an energy level • number of orbital shapes possible in energy level = n Orbital Shapes designated s,p ...

note_ep304aditi.pdf - orientation of orbital in space - 0

Note Taking Guide: Episode 304 Name _____ Quantum Numbers n, l, m, s Used to describe an electron in an atom _____ n principle quantum _____ number Represents main energy level of _____electron maximum # of _____electron _____ in an energy level _____ = 2n² _____ Example: What is the maximum number of electrons that can be in the 5th _____ main energy level? l The 2 nd quantum number _____ Describes the orbital _____shape within an _____

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