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WORKSHEET Ionic Bonds Use
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1. How many valence
electrons does a neutral
magnesium (Mg) atom have? 2.
What is the charge on a
magnesium ion? What does

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magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl) atom have? 4.

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11 Name: Date: Hour:

Information: Energy Levels

and Sublevels ...

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question 5, how many
electrons can fit in the
entire 5t energy level? can
(AVel is a ha AC5a) 50đ 7.
Write the notation for an
electron spinning clockwise
in a p sublevel ...

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WORKSHEET The Activity

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Section 9.2 1. For each of

the following pairs of

elements, underline the one

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that would replace the other
element in a compound. a.
calcium, tin e. iron, copper
b. bromine, fluorine f.
iodine, chlorine c.
aluminum, potassium g.
silver, lead d. zinc, sodium
2.

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WORKSHEET The Activity

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1.7, the CF bond is polar

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covalent. (b) $\Delta EN_{\text{PCl}} = EN_{\text{Cl}} - EN_{\text{P}} = 3.0 - 2.1 = 0.9$. Since $0.5 < \Delta EN < 1.7$, the PCl bond is polar covalent. (c) $\Delta EN_{\text{LiCl}} = EN_{\text{Cl}} - EN_{\text{Li}} = 3.0 - 0.9 = 2.1$. Since $\Delta EN > 1.7$, the LiCl bond is ionic.

Section 8.5

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contrast polar that is in
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What is the shape of an s
orbital? Class 16 Use with
Chapter 5, Section 5.2 2.

What is the relationship

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Chapter 19
between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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